

Optimization of resveratrol dimerization in a batch reactor using design of experiments methodology

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21 Ružičkine dani
DANAS ZNANOST – SUTRA INDUSTRIJA
9. – 11. rujna 2026. • Vukovar • Hrvatska

INTRODUCTION

Resveratrol is a bioactive stilbene whose oxidative and photochemical coupling can generate structurally diverse dimers with potentially improved stability and biological activity. However, competing isomerization, oxidation, and oligomerization often limit dimer yield and selectivity.

In this study:

Photochemical dimerization under UV irradiation and FeCl₃-mediated oxidative coupling were systematically investigated in a batch reactor.

MATERIALS AND METHODS

Design of experiments (DoE) was used to identify significant process factors and operating conditions that favour selective resveratrol dimerization.

Experimental design

Photochemical dimerization

Table 1a. Factor levels used for optimization of resveratrol dimerization under UV irradiation.

Condition	Low level (-1)	Centre level (0)	High level (+1)
<i>t</i> (h)	1	2.5	4
<i>n</i> _{resveratrol} (mol)	0.00001	0.0001	0.001
Number of lamps (-)	8	12	16

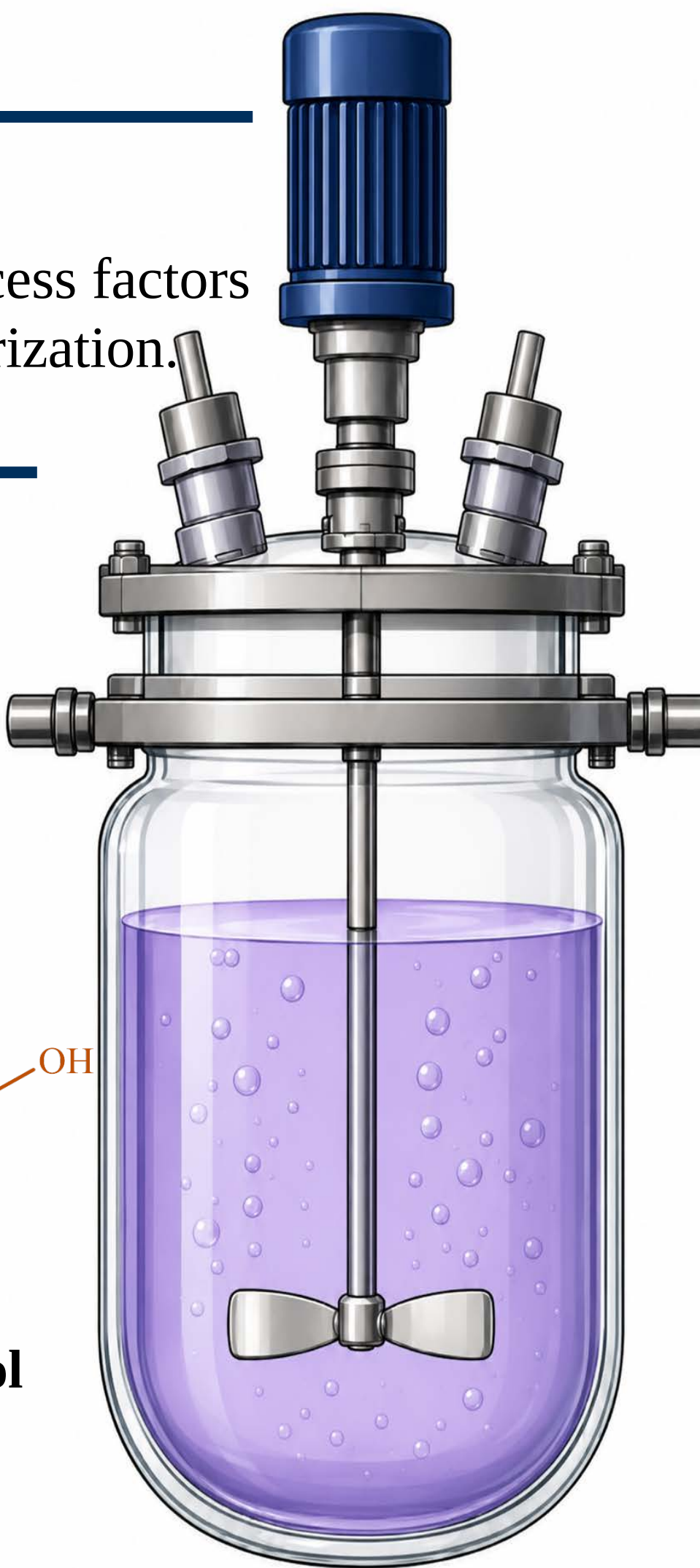
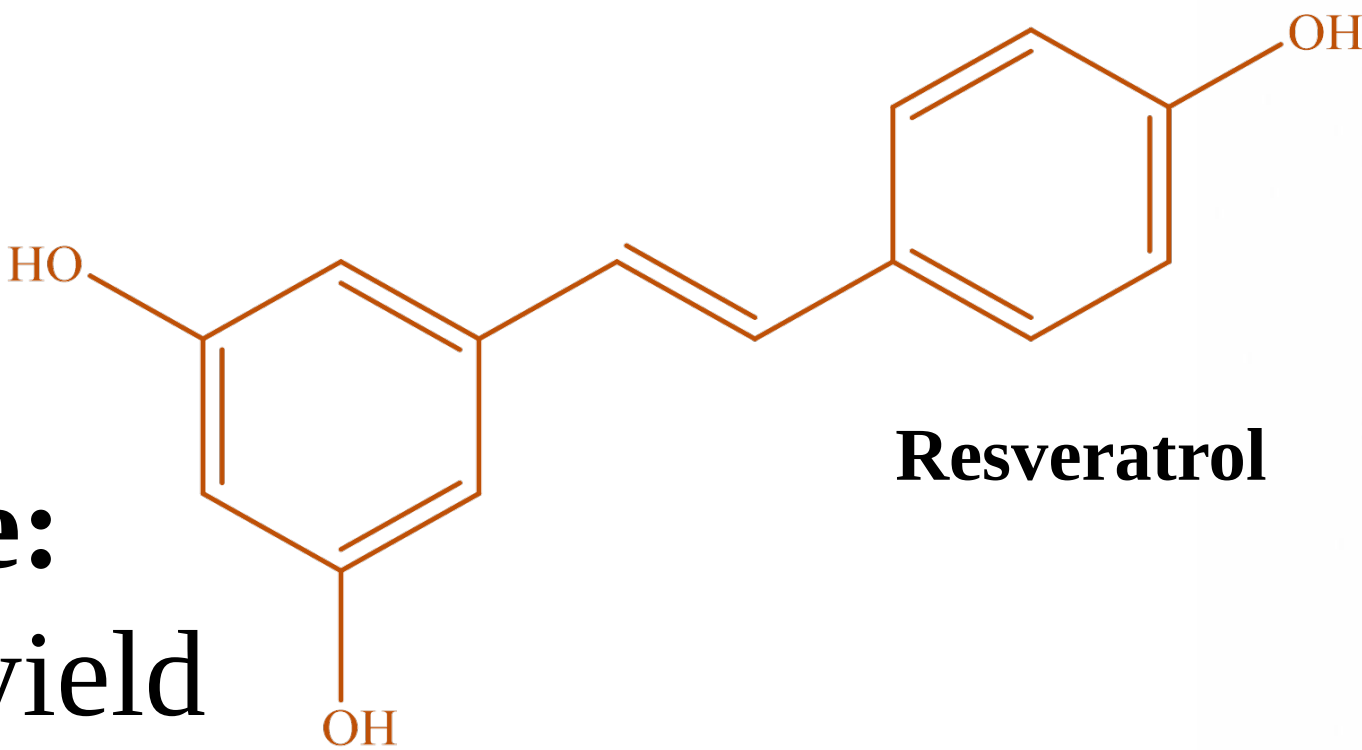
Oxidative coupling

Table 2a. Factor levels used for optimization of FeCl₃-mediated oxidative coupling.

Condition	Low level (-1)	Centre level (0)	High level (+1)
<i>c</i> _{resveratrol} (mmol/L)	5	15	25
FeCl ₃ (eq)	0.50	1.25	2.00
<i>T</i> (°C)	20	30	40

Two independent Box–Behnken designs with centre-point replicates were applied (Table 1a and 2a). Photochemical factors were irradiation time, resveratrol amount, and number of UV lamps; oxidative-coupling factors were resveratrol concentration, FeCl₃ amount, and temperature.

Response:
δ-Viniferin yield



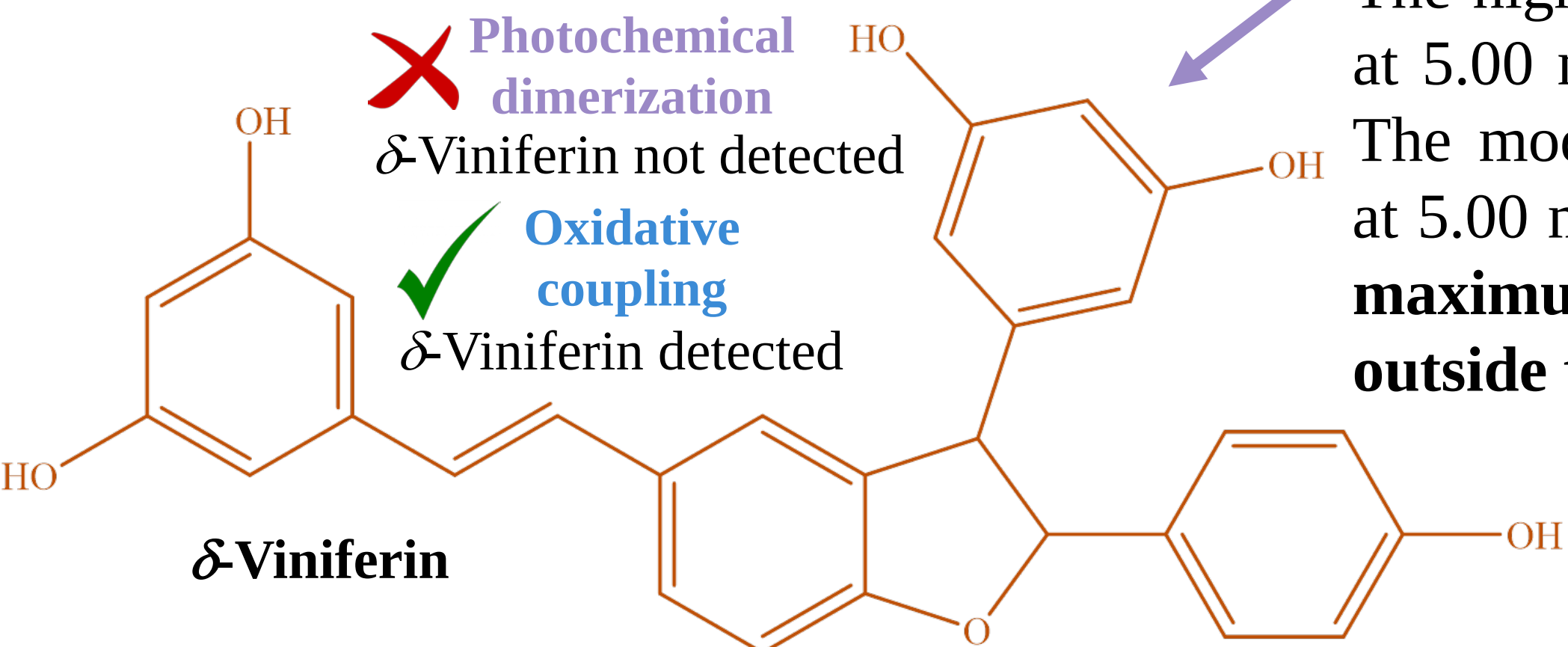
RESULTS

Direct UV irradiation of unprotected resveratrol did not produce the targeted dimer under any investigated condition (Table 1b; yield: 0%). The result agrees with reports that UV exposure primarily promotes *trans/cis*-isomerization and singlet-oxygen-related oxidation rather than intermolecular dimerization (DOI: 10.1039/C2OB26241J; 10.1071/CH20176).

Varying substrate amount, irradiation time, and lamp number was therefore insufficient.

Table 1b. Box–Behnken experimental design and experimentally obtained product yield.

Experiment	<i>t</i> (h)	<i>n</i> _{resveratrol} (mol)	Number of lamps (-)	Yield (%)
1	1.0	0.001000	12	0.00
2	4.0	0.001000	12	0.00
3	1.0	0.000001	12	0.00
4	4.0	0.000001	12	0.00
5	1.0	0.000100	8	0.00
6	4.0	0.000100	8	0.00
7	1.0	0.000100	16	0.00
8	4.0	0.000100	16	0.00
9	2.5	0.001000	8	0.00
10	2.5	0.000001	8	0.00
11	2.5	0.001000	16	0.00
12	2.5	0.000001	16	0.00
13	2.5	0.000100	12	0.00
14	2.5	0.000100	12	0.00
15	2.5	0.000100	12	0.00
16	2.5	0.000100	12	0.00
17	2.5	0.000100	12	0.00



CONCLUSION

Direct UV irradiation did not produce the targeted dimer, whereas FeCl₃-mediated oxidative coupling achieved a maximum experimental yield of 16.84%. DoE identified a boundary optimum that requires validation....**back to the drawing board!**

FeCl₃ was selected as a **mild, inexpensive, and readily available oxidant** capable of generating phenoxy radicals and promoting oxidative C–C coupling.

Table 2b. Box–Behnken experimental design and experimentally obtained product yield.

Experiment	<i>c</i> _{resveratrol} (mmol/L)	FeCl ₃ (eq)	<i>T</i> (°C)	Yield (%)
1	5.00	0.50	30.00	12.42
2	25.00	0.50	30.00	3.15
3	5.00	2.00	30.00	5.46
4	25.00	2.00	30.00	0.27
5	5.00	1.25	20.00	5.84
6	25.00	1.25	20.00	1.92
7	5.00	1.25	40.00	16.84
8	25.00	1.25	40.00	0.40
9	15.00	0.50	20.00	4.75
10	15.00	2.00	20.00	0.18
11	15.00	2.00	40.00	0.50
12	15.00	0.50	40.00	2.54
13	15.00	1.25	30.00	2.59
14	15.00	1.25	30.00	1.50
15	15.00	1.25	30.00	0.00
16	15.00	1.25	30.00	0.94
17	15.00	1.25	30.00	1.91

Future studies:

- protect hydroxyl groups,
- test solid-state or host-guest irradiation,
- use an inert atmosphere or flow photochemistry.

ANOVA results

Significant (*p* < 0.05)

- *c*_{resveratrol}: *p* = 0.00036
- FeCl₃ amount: *p* = 0.0191
- Interaction between resveratrol concentration and temperature: *p* = 0.0138

The highest experimental yield was 16.84% (Table 2b) at 5.00 mmol/L resveratrol, 1.25 eq FeCl₃, and 40 °C. The model predicted a boundary optimum of 16.96% at 5.00 mmol/L, 0.50 eq FeCl₃, and 40 °C. **No internal maximum was identified; the stationary point outside the investigated range was a saddle point.**

Future studies should:

- validate the predicted optimum,
- expand the experimental range, and
- screen Cu(II) and Mn(III) salts, laccases, peroxidases, and cytochrome P450 enzyme

