

SORPTION OF COPPER IONS ON IMMOBILISED ZEOLITE AND YEAST

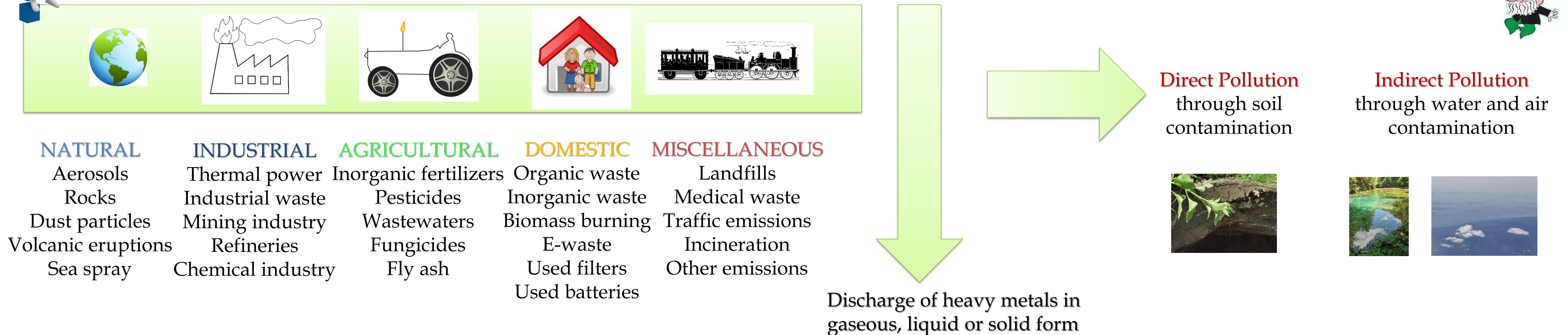
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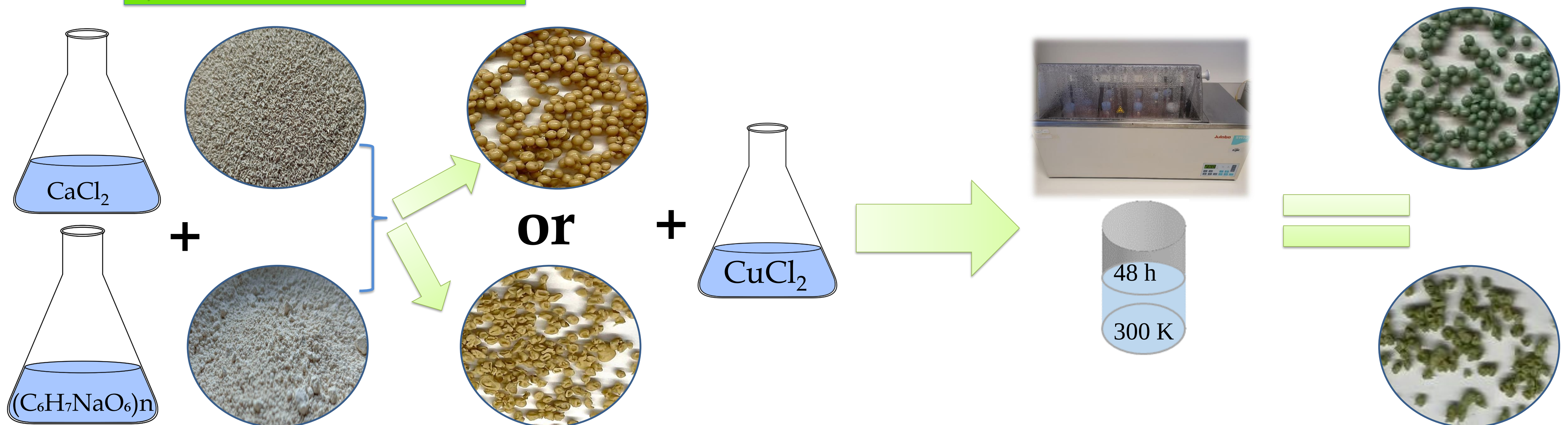
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1) INTRODUCTION

An alarming environmental problem is heavy metal ions pollution.



2) EXPERIMENTAL PART



3) RESULTS

Parameter	Sorbent incorporated in alginate		Parameter	Sorbent incorporated in alginate	
	zeolite NaX	Yeast <i>Saccharomyces cerevisiae</i>		zeolite NaX	Yeast <i>Saccharomyces cerevisiae</i>
$q_{\text{max},e}$ (mmol/g)	0.437	0.065	$q_{\text{max},e}$ (mmol/g)	0.437	0.079
Freundlich isotherm			Langmuir isotherm		
K (mmol/g)	0.268	0.025	K (dm ³ /mmol)	1.153	0.228
n	0.277	0.482	M (mmol/g)	0.494	0.105
RMSE	0.022	0.014	RMSE	0.016	0.013
χ^2	0.007	0.019	χ^2	0.005	0.016
Redlich-Peterson isotherm			Sips isotherm		
K (dm ³ /mmol) ^{β}	1.528	4.743	K (dm ³ /mmol) ^{β}	0.961	107.879
M (mmol/g)	0.433	0.018	M (mmol/g)	0.538	0.067
β	0.948	0.603	β	0.817	0.213
RMSE	0.019	0.014	RMSE	0.016	0.019
χ^2	0.007	0.019	χ^2	0.004	0.035

	Sorbent incorporated in alginate				
	Zeolite NaX		Yeast <i>Saccharomyces cerevisiae</i>		
c_0 mmol/dm ³	q_e mmol/g	R %	q_e mmol/g	R %	
2.165	0.226	70.21	0.025	7.67	
3.334	0.279	56.96	0.034	6.83	
5.249	0.387	50.41	0.079	10.34	
7.010	0.410	39.55	0.059	5.71	
9.825	0.437	30.29	0.065	4.51	

4) CONCLUSIONS

The copper sorption data on immobilised NaX and yeast was best described with the Langmuir isotherm model. The FTIR spectra of the sorbents used, recorded prior and after the sorption, revealed no changes in the structure of the sorbents.

