



PRECIPITATION OF THE UNSTABLE HYDRATED FORM OF CALCIUM OXALATE

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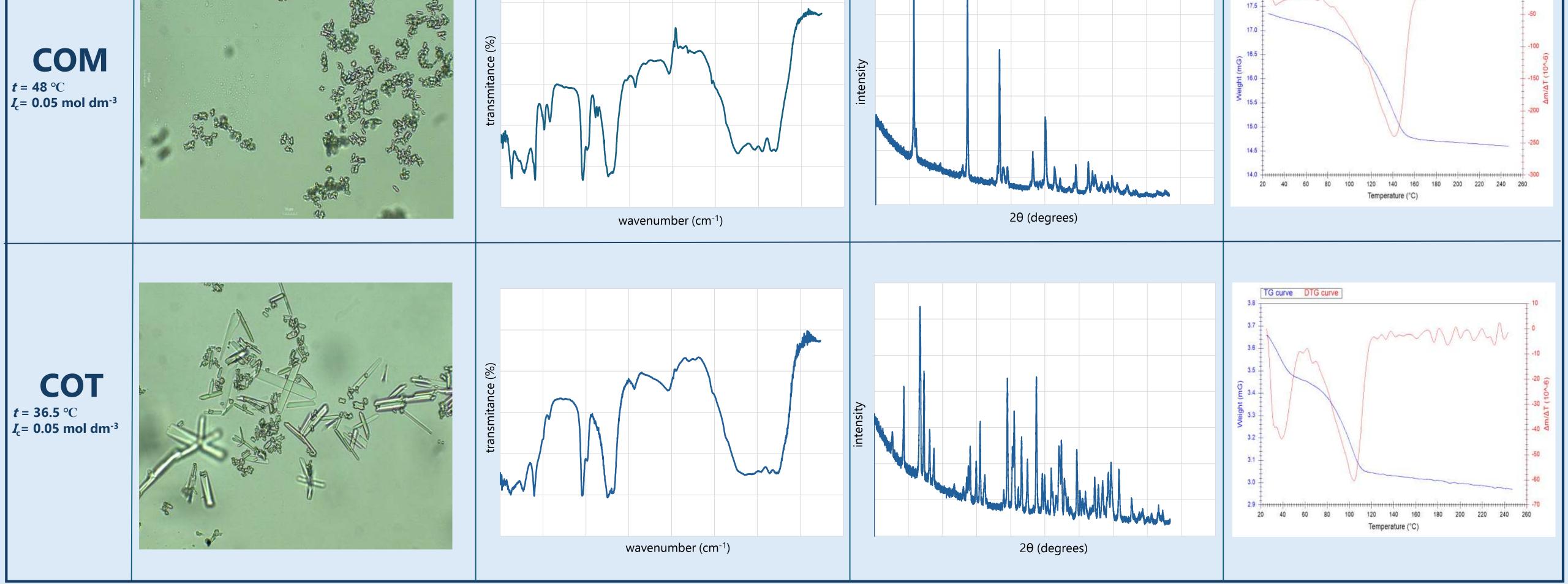
INTRODUCTION

Pathological biomineralization is the process of the formation of undesirable biominerals, e.g. kidney stones. Their formation causes a chronic disease - urolithiasis. Kidney stones occur more and more frequently nowadays due to a fast-paced life and an unbalanced diet. The composition of kidney stones is calcium oxalate in 80% of cases. Calcium oxalate occurs in 3 different forms: the stable form calcium oxalate monohydrate (COM) and the unstable forms calcium oxalate dihydrate (COD) and calcium oxalate trihydrate (COT). In the composition of kidney stones, COM and COD are most frequently present, while COT is rare. This work aims to synthesize COT and to study the morphology of the obtained crystals at three temperatures (25, 36.5 and 48 °C) and three ionic strengths (0.1, 0.05 and 0.1 mol dm⁻³).

	ĩ	Conducted		Temperature (°C)		
6,533 pH	synthesis		25 °C	36.5 °C	48 °C	
	6,533 pH	Ionic	0.01	COT	COT	COM
		strength (I _c /mol dm ⁻³)	0.05	COT	COT	COM
		$(I_c/1101 \text{ dm}^2)$	0.1	COT	COT	COM

RESULTS

ΤΥΡΕ	MICROSCOPE	FTIR	XRD	TGA
				18.0 TG curve DTG curve



CONCLUSIONS

Analysis of the synthesized crystals by optical microscopy, Fourier transform infrared spectroscopy (FTIR), X-ray powder diffraction (XRD) and thermogravimetric analysis (TGA) shows that COT is formed at temperatures of 25 and 36.5 °C. In contrast, COM is formed at 48 °C. For all systems, the ionic strength was shown to have no effect on the formation of crystals or their appearance.

